



Electron density profiles in the quiet lower ionosphere based on the results of modeling and experimental data

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Received: 23 August 2011 – Revised: 16 August 2012 – Accepted: 16 August 2012 – Published: 13 September 2012

Abstract. The theoretical PGI (Polar Geophysical Institute) model for the quiet lower ionosphere has been applied for computing the ionization rate and electron density profiles in the summer and winter D-region at solar zenith angles less than 80° and larger than 99° under steady state conditions. In order to minimize possible errors in estimation of ionization rates provided by solar electromagnetic radiation and to obtain the most exact values of electron density, each wavelength range of the solar spectrum has been divided into several intervals and the relations between the solar radiation intensity at these wavelengths and the solar activity index $F_{10.7}$ have been incorporated into the model. Influence of minor neutral species (NO, H₂O, O, O₃) concentrations on the electron number density at different altitudes of the sunlit quiet D-region has been examined. The results demonstrate that at altitudes above 70 km, the modeled electron density is most sensitive to variations of nitric oxide concentration. Changes of water vapor concentration in the whole altitude range of the mesosphere influence the electron density only in the narrow height interval 73–85 km. The effect of the change of atomic oxygen and ozone concentration is the least significant and takes place only below 70 km.

Model responses to changes of the solar zenith angle, solar activity (low–high) and season (summer–winter) have been considered. Modeled electron density profiles have been evaluated by comparison with experimental profiles available from the rocket measurements for the same conditions. It is demonstrated that the theoretical model for the quiet lower ionosphere is quite effective in describing variations in ionization rate, electron number density and effective recombination coefficient as functions of solar zenith angle, solar activity and season. The model may be used for solving inverse

tasks, in particular, for estimations of nitric oxide concentration in the mesosphere.

Keywords. Ionosphere (Ion chemistry and composition; Ionization mechanisms; Modeling and forecasting)

1 Introduction

Characteristic features of the ion structure of the D-region between 50–90 km heights are the presence of ion clusters, formation of simple and complex negative ions, interactions between primary ions and chemically active minor neutral components. Knowledge of the ionospheric structure, in particular the height distribution of the electron number density as a function of geographical location, season and time is important for radio wave propagation. From a scientific point of view, investigation of the lower ionosphere and examination of its structure allow the study of coupling processes between the neutral atmosphere and the ionospheric plasma, and between space weather and Earth's geospace.

Despite of development and improvement of radio-physical methods for measurements of the electron number density, including ground-based techniques and sounding rockets, there is a large uncertainty in the obtained results. Moreover, during quiet conditions and weak auroral events, experimental data on electron density at altitudes below 80 km are often absent. Incoherent scattering of radio waves gives reliable information about the height distribution of the electron number density during quiet periods at altitudes above 75–80 km. The partial reflection method allows reliable electron number density to be obtained in the quiet ionosphere only at altitudes above 80 km. Rocket

measurements that apply the Faraday rotation method give the most reliable data, but do not provide long-term data sets. In order to improve methods for prediction of radio wave propagation and to ensure communication, better knowledge of the spatial and temporal distribution of the electron density under various conditions is needed, i.e. in geomagnetic quiet and disturbed situations, for different seasons and solar zenith angles. There are several empirical and semi-empirical models of the D-region (for example, Friedrich and Torkar, 1992, 1998, 2001; Danilov, 2000; McKinnell and Friedrich, 2007) that have been proposed as inputs to the International Reference Ionosphere model (IRI). However, these models have not been applied for description of the great variability in the electron number density below 75–80 km nor under specific geophysical situations. This is especially important for stable operation, navigation and communications in very low and low frequency (VLF/LF) ranges, since these waves are reflected from the lower border of the D-region.

Theoretical models are useful tools for examining the structure of the lower ionosphere during various geophysical situations and at any heights if they take into account all basic processes and factors that influence the ionization balance. The principal purpose and the complexity of theoretical model construction is the correct description of the physical and chemical processes that create and support the ionosphere in the D-region. These processes are ionization and different types of chemical and recombination reactions. Efficiencies of ion-molecular reactions which, together with processes of ionization and recombination, form ionosphere structures depend on the local concentrations of long-lived species such as NO, H₂O, O and O₃. They, in turn, depend on solar illumination, season, ionization sources, solar activity, turbulence, diffusion, transport and other factors. Therefore, it is important to incorporate, into any model, realistic height profiles of the minor constituents that are appropriate for a given geophysical situations. However, for the present there is a lack of accurate experimental information on the height distribution of many of these constituents. Measurements demonstrate a significant difference in minor species concentrations at the altitudes of the mesosphere for similar geophysical conditions. Theoretical estimations of numerical density of minor species, obtained with different diffusive-photochemical models, but for the same conditions, also differ from each other and are not in good correlation with the experimental data. This creates a problem for suitable choice of minor constituents profiles in each concrete case and can lead to inaccuracies in the modeling of electron density and other ionospheric parameters.

The ion-chemical model developed by Smirnova et al. (1988) for the lower ionosphere has been used by Osepian et al. (2008, 2009a, b) to investigate the influence of the atomic oxygen and ozone concentrations on the electron density and the ion composition at different altitudes in the D-region during solar proton events, when there are favorable conditions for accurate measurements of electron den-

sity in the whole D-region, including the lower altitudes. The total PGI (Polar Geophysical Institute) model of ionization-recombination cycle for the D-region takes into account sources of ionization, ion-molecular reactions that form positive cluster ions and complex negative ions from the primary ions, their dependence on temperature and number densities of minor neutral constituents, electron attachment to neutrals, electron photo-detachment of negative ions, photodissociation of negative ions, recombination of positive ions with electrons, reactions of ion-ion recombination. A computer algorithm for computation of ionization rates produced by solar protons has been developed and tested by Kirkwood (1988) and Kirkwood and Osepian (1995). Reliable measurements of solar proton fluxes in several energy intervals (by the satellites GOES) permit good estimations of ionization rates and electron number densities. Atomic oxygen and ozone profiles which lead to model electron number density profiles $N_e(h)$ consistent with observations and which, consequently, are the most appropriate for polar mesospheric altitudes under solar proton event conditions have been presented by Osepian et al. (2008, 2009a).

In this study, the theoretical PGI model of the D-region has been modified to examine the structure of the lower ionosphere during quiet conditions and at any geographic latitude. In order to minimize possible errors in estimation of ionization rates by solar electromagnetic radiation which might be neglected during solar proton precipitation, a new, more accurate algorithm for computation of solar radiation intensity at different levels of solar activity has been applied. For this aim the dependences of the solar radiation intensity on the solar index $F_{10.7}$ derived by Nusinov (1992) for each wavelength range of the solar spectrum were used. The model of the lower ionosphere consists of the following modules:

- model of the neutral atmosphere's major constituents (MSIS-2000) (Hedin, 1991; Picone et al., 2002);
- height profiles of the minor neutral constituents;
- algorithm for calculating solar radiation intensity in different wavelength ranges;
- algorithm for calculating the ionization rates taking into account all sources of the ionization during quiet periods;
- algorithm for calculating concentration of minor neutral constituents for different solar zenith angles and different altitudes;
- ion chemistry model (Smirnova et al., 1988).

This type of model construction provides good opportunities for model modification when new data on ionospheric and mesospheric parameters, ionization sources and chemical processes appear.

The ion chemical model, implemented as a separate module in the theoretical model of ionization-recombination cycle for D-region, has been developed by Smirnova et al. (1988). In this model the simple cluster ions $\text{NO}^+(\text{H}_2\text{O})_n$ and $\text{O}_2^+(\text{H}_2\text{O})_n$, formed from the primary ions NO^+ and O_2^+ , are combined into a cluster family 1, Cl_1^+ . Their recombination coefficient with electrons is typically $2 \times 10^{-6} \text{ cm}^3 \text{ s}^{-1}$. Proton hydrates $\text{H}^+(\text{H}_2\text{O})_n$, are combined into a cluster family 2, Cl_2^+ . They recombine with electrons about five times faster than ions from Cl_1^+ . The primary negative ions O_2^- and O^- initiate series of reactions with neutral constituents and produce complex negative ions, in particular O_3^- , O_4^- , CO_2^- , CO_4^- . These ions are intermediate, have short life time and low concentrations. However, their role in formation of the major intermediate ion CO_3^- and the stable NO_3^- ion, which is formed from CO_3^- , is very important.

In the ion-chemical model that was used in this study, the efficiency of the hydration chains of NO^+ and O_2^+ and reaction chains forming intermediate negative ions CO_3^- from the primary ions O_2^- and O^- are substituted by the effective parameters $B(\text{N}^{+})$, $B(\text{O}_2^+)$, $B(\text{O}_2^-)$ and $B(\text{O}^-)$ which are determined by the expressions (A1)–(A4) given in the Appendix. The transformation of Cl_1^+ to proton hydrates $\text{H}^+(\text{H}_2\text{O})_n$ is described by the expression (A5). The rate constants for positive and negative ion reactions are given in Tables A1 and A2, respectively.

The input parameters for the model are date, local time, geomagnetic latitude, geographic latitude and longitude, temperature, altitude profiles for O_2 , N_2 , height profiles for minor neutral constituents (NO , O , O_3 , H_2O , $\text{O}_2(1\Delta_g)$, NO_2 , CO_2) for noon and midnight LT with a vertical step of 1 km and solar activity index $F_{10.7}$. Due to the short lifetime of ions and electrons in the D-region, dynamical and transport effects are ignored. The continuity equations for the main positive ions (NO^+ , O_2^+ , Cl_1^+ , Cl_2^+) and negative ions (O_2^- , O^- , CO_3^- , NO_3^-) are solved under steady-state conditions. Electron number density N_e is calculated as the difference between positive and negative ion number densities. Under equilibrium conditions, effective recombination coefficient α_{eff} is determined from the continuity equation for electrons $Q(h) = \alpha_{\text{eff}} \cdot N_e^2$. More detailed description of the ion-chemical model is given by Smirnova et al. (1988), Osepian et al. (2008).

The aim of this study is

- to model electron density profiles in the quiet lower ionosphere using NO , O , O_3 , and H_2O profiles with different values of concentration and evaluate their influence on the electron number density at the different altitudes;
- to examine the response of the model to changes of the solar zenith angle, solar activity and season;

- to verify the modeled $N_e(h)$ -profiles with experimental data in the sunlit summer and winter ionosphere and to determine which profiles of minor neutral constituents lead to the best agreement between theoretical and experimental N_e values for chosen situations.

A brief description of the ionisation sources in quiescent conditions is presented in Sect. 2. The height profiles of the mesospheric minor neutral species (NO , O , O_3 , H_2O) are given in Sect. 3. Model simulations of the ionospheric parameters in the daytime summer and winter ionosphere are given in Sect. 4. Section 5 contains discussion and conclusions.

2 Sources of ionization in the quiescent conditions

2.1 Daytime conditions

Solar radiation is the leading source of ion production in the daytime ionosphere under quiet geophysical conditions. Ionization of the D-region during the daytime is determined by several wavelength ranges, namely, hard X-rays (less than 1 nm), soft X rays (1–10 nm), Ly- β (102.6 nm) and Ly- α (121.6 nm) emissions, the continuum radiation at 102.6–111.8 nm. The ion production rate at the altitude h is determined by the classic formula:

$$Q_{\text{ph}}(h) = \sum_{\lambda} J_{\infty}(\lambda) \sum_i \sigma_i(\lambda) n_i(h) \exp \left\{ - \sum_i \bar{\sigma}_i(\lambda) \int_h^{\infty} N_i(z) Ch(\chi) dz \right\} \quad (1)$$

where $J_{\infty}(\lambda)$ is the intensity of the solar radiation on the top of the atmosphere at wavelength λ , $\sigma_i(\lambda)$ is the ionization cross section, $\bar{\sigma}_i(\lambda)$ is the absorption cross section, $n_i(h)$ is the concentration of the neutral component at the altitude h and $\int_h^{\infty} N_i(z) Ch(\chi) dz$ is the content of absorbing component in the atmosphere column above the given altitude h , $Ch(\chi)$ is the Chapman grazing incidence function, which can be calculated according to Smith and Smith (1972). Ionization and absorption cross sections have been evaluated for different wavelengths by Nicolet and Aikin (1960); Ivanov-Kholodniy and Velichanskiy (1973); Torr et al. (1979); Rees (1989); Nusinov (1986, 1992).

For each wavelength range of the solar spectrum, solar radiation intensity has a different dependence on the level of solar activity where the latter can be described by the parameter $F_{10.7}$. For calculation of solar X-ray and EUV-radiation intensities as functions of solar activity a method developed by Nusinov (1992) has been applied. According to this method, solar radiation intensity in the wavelength range 10–102.6 nm can be calculated using the expression:

$$J_{\lambda}(F_{10.7}) = R_{\lambda} \times J_{58.4}(F_{10.7}) \quad (2)$$

where the wavelength of 58.4 nm is assumed to be a principal line. The value R_λ for every line is calculated using the following expression with coefficients a_λ and b_λ as given by Nusinov (1992):

$$R_\lambda = a_\lambda + b_\lambda J_{58.4}(F_{10.7}). \quad (3)$$

The order and units of $J_{58.4}(F_{10.7})$ are $10^9 \text{ cm}^{-2} \text{ s}^{-1}$. It depends on the solar activity according to the expression:

$$J_{58.4}(F_{10.7}) = 1.38 + 0.111(F_b - 60)^{0.667} + 0.0538(F_{10.7} - F_b)^{0.667}. \quad (4)$$

The photoionization and photo-absorption cross-sections for O_2 , N_2 and O within the wavelength range 10–102.6 nm are taken according to Torr et al. (1979) and Rees (1989).

Variation of the Ly- α intensity as a function of solar activity is given by the expression:

$$J_{121.6}(F_{10.7}) = \{134 + 39.9(F_b - 60)^{0.667} + 5.47(F_{10.7} - F_b)^{0.667}\} \times 10^9 \quad (5)$$

where, F_b is the background flux of the solar radio emission at the wavelength of 10.7 cm. Variations of F_b during the solar cycle are defined as

$$F_b = \sum_{i=0}^4 [A_i \cos(2\pi it/T) + B_i \sin(2\pi it/T)] \quad (6)$$

with the coefficients A_i (82.1, -19.6, 1.778, 2.59, -2.33) and B_i (0, 10.55, -0.956, 3.104, -0.925) incorporated into the program from Nusinov (1992). T is the duration of the solar cycle, t is the number of years from the solar cycle minimum.

The X-ray intensity for varying solar activity conditions can be expressed as

$$J_{\lambda_1-\lambda_2}(F_{10.7}) = J_{\lambda_1-\lambda_2}^0 [I_r(F_{10.7})/16.8]^d \quad (7)$$

with parameter d determined as

$$d = 0.22 + 1.56/\lambda_2. \quad (8)$$

In Eq. (7) $I_r(F_{10.7})$ is the energy flux given in $\mu\text{W m}^{-2}$ within the wavelength interval 0.8–2.0 nm. $I_r(F_{10.7})$ varies with the solar activity according to:

$$I_r(F_{10.7}) = 0.29F_{10.7} - 18 \quad (9)$$

The values of $J_{\lambda_1-\lambda_2}^0$ and the absorption and ionization cross-sections for all intervals of the X-ray spectrum are given by Nusinov (1986, 1992). The intensity of the continuum radiation within the wavelength range 102.7–111.8 nm depends on the solar activity according to Paulsen et al. (1972).

2.2 Nocturnal conditions

For calculation of the ion-production rate in nocturnal conditions, a semi-empirical approach developed by Kashirin (1986) has been applied. This approach is based on the combined analysis of theoretical and experimental data on spatial and time variations of the scattered radiation and provides the total isotropic flux at an altitude of 300 km for four wavelengths, i.e. 30.4 nm (He II), 58.4 nm (He I), 121.6 nm (Ly- α), and 102.6 nm (Ly- β). The flux is a function of solar zenith angle χ , solar activity $F_{10.7}$ and season. The ion production rate due to scattered radiation at wavelength λ at altitudes below 300 km is given by

$$q^\lambda = \sigma_\lambda^i n_i \int_0^{2\pi} d\phi \int_0^\pi \sin\theta d\theta J_\lambda(\theta, \phi) \exp[-\tau_\lambda \sec\theta] \quad (10)$$

where

$$\tau_\lambda = \sigma_\lambda^a \int_h^{h_{\max}} [\text{O}_2](h) dh. \quad (11)$$

In Eqs. (10) and (11) σ_λ^i is the ionization cross section at wavelength λ , σ_λ^a is the absorption cross section, θ is the zenith angle, ϕ is the azimuth angle, n_i is the concentration of the component that absorbs and scatters the radiation at the characteristic altitude h , i.e. NO for Ly- α radiation, O_2 for Ly- β line, O_2 , N_2 and O for He I and He II lines. The scattered radiation at 300 km is assumed to be isotropic over the downward hemisphere $J_\lambda(\theta, \phi) = J_\lambda$.

The model takes into account the ionization by galactic cosmic rays, whose intensity depends on geomagnetic latitude and solar activity, $F_{10.7}$, according to Swider (1979).

3 The minor neutral constituents

Ion composition and molecular reactions in the D-region are directly coupled to the local concentration of the long-lived minor neutral species including NO, H_2O , O and O_3 . Nitric oxide, NO, is the most important component for the D-region ion chemistry under quiet conditions. Ions NO^+ are the main source of cluster ions in the quiet lower ionosphere. Molecules of NO actively participate in complex photochemical reactions, resulting in formation of the final ions NO_3^- . Experimental data demonstrate the dependence of NO number density on latitude, solar activity and season (Grossmann et al., 1985; Siskind and Rusch, 1992; Siskind et al., 1997, 1998). Theoretical studies have revealed the processes controlling photochemical sources and sinks of nitric oxide (Ogawa and Shimazaki, 1975; Siskind et al., 1997, 1998). The reactions that are responsible for production of odd nitrogen, i.e. nitrogen oxide and atomic nitrogen, in the lower thermosphere, are summarized in Table 1.

Table 1. Reactions responsible for odd nitrogen production (Ogawa and Shimazaki, 1975; Marov et al., 1996; DeMore et al., 1994; Sander et al., 2003).

No.	Reaction	Rate constant, cm ³ s ⁻¹
(R1)	N ₂ + hν (800–1000 Å) → N + N	1.90 × 10 ⁻⁸
(R2)	N ₂ + hν (1–800 Å) → N + N	4.53 × 10 ⁻⁸
(R3)	N ₂ O + hν (630–2490 Å) → NO + N	3.71 × 10 ⁻⁸
(R4)	NO ₂ + hν (2440–3980 Å) → NO + O	8.30 × 10 ⁻³
(R5)	NO ₂ + hν (1027–2440 Å) → NO + O(¹ D)	1.22 × 10 ⁻⁴
(R6)	N ₂ + p ⁺ → N(² D) + N(⁴ S) + p ⁺	–
(R7)	N ₂ + e → N(² D) + N(⁴ S) + e	–
(R8)	N ₂ ⁺ + O → N(² D) + NO ⁺	1.4 × 10 ⁻¹⁰ · (300/T) ^{0.44}
(R9)	NO ⁺ + e → N(² D) + O	4.2 × 10 ⁻⁷ (300/T) ^{0.85}
(R10)	N(² D) + O ₂ → NO + O	1.5 × 10 ⁻¹¹ · exp(-3600/T)

p⁺ is primary proton, e is secondary electron.

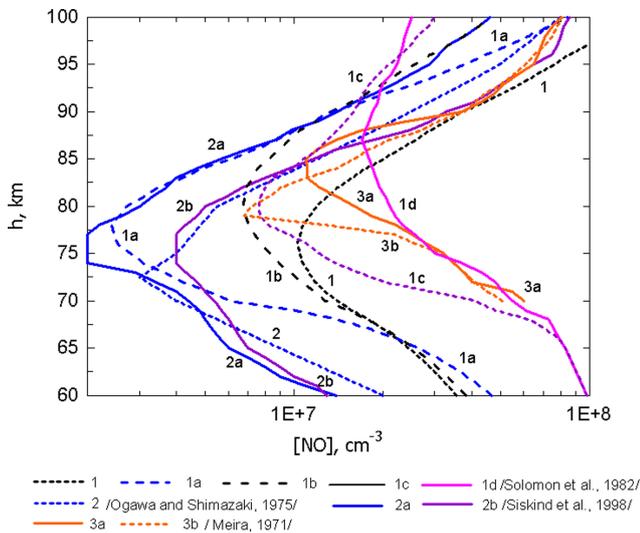


Fig. 1a. Examples of nitric oxide number density profiles during the daytime (solar zenith angle below 90°) retrieved from photochemical models 1, 1a, 1b, 1c, 1d (Solomon et al., 1982) and experimental measurements 2 (Ogawa and Shimazaki, 1975), 2a, 2b (Siskind et al., 1998), 3a, 3b (Meira, 1971).

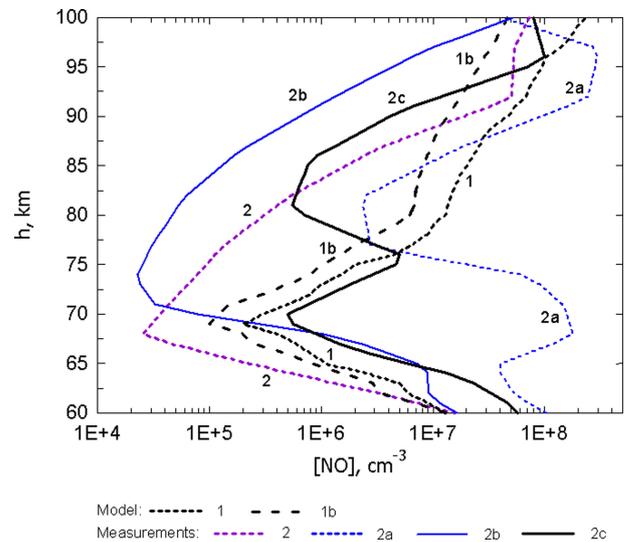


Fig. 1b. Examples of nitric oxide number density profiles during the nighttime (solar zenith angle above 94°) retrieved from photochemical models 1, 1b (Solomon et al., 1982) and measurements 2, 2a, 2b, 2c (HALOE <http://haloe.gats-inc.com/home/index.php>).

Reactions (R1)–(R5) are responsible for photo dissociation of N₂, N₂O and NO₂ in quiet conditions. At high latitudes, odd nitrogen can be produced by the collisions between nitrogen and relativistic protons (Reaction R6) or secondary electrons (Reaction R7). According to Porter et al. (1976), during the Reaction (R6), 45 % of nitrogen atoms are in the state [N(⁴S)] and 55 % are in the excited state [N(²D)]. Since there is a strong coupling between the neutral and ionized species, the reactions involving ionic species (Reactions R8, R9) are important to increase the concentration of NO through the Reaction (R10). Downward transport from the thermosphere is the main source of NO in the mesosphere. This has been confirmed by Halogen Occultation

experiment (HALOE) on board the Upper Atmosphere Research Satellite (UARS) which has demonstrated significant coupling between the NO layers in the thermosphere and the middle atmosphere (Siskind and Russel, 1996; Siskind et al., 1998). Examples of NO profiles, including available profiles with minimum and maximum NO concentrations, are shown for daytime and nighttime in Fig. 1a and b. The discrepancies in the numerical density for daytime NO profiles (Fig. 1a) might be about one order of magnitude. Difference in the nighttime NO concentrations (Fig. 1b) might be a few orders of magnitude. The nighttime profiles 1 and 1b (Fig. 1b) have been calculated from the respective daytime profiles 1 and 1a (Fig. 1a) using the coefficients from Ogawa and Shimazaki (1975). The NO profiles 2, 2a, 2b and

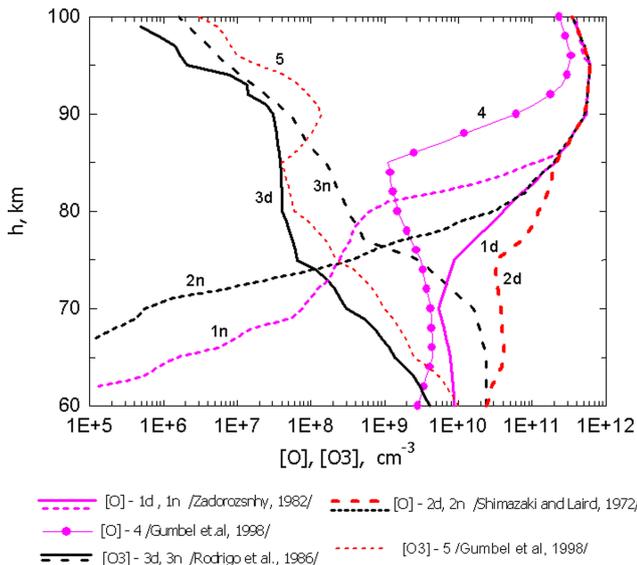


Fig. 2a. Profiles of atomic oxygen and ozone in the sunlit and nocturnal mesosphere (d – day; n – night).

2c have been obtained from the HALOE instrument (2011): (<http://haloe.gats-inc.com/home/index.php>). All profiles of nitric oxide presented in Fig. 1a and b have been used as inputs for modeling the electron density profiles.

Modern knowledge on the height distribution of atomic oxygen in the quiet mesosphere, especially below 80 km, is mainly based on the results of the theoretical estimations derived from diffusion-photochemical models (Shimazaki and Laird, 1970, 1972; Turco and Sechrist, 1972; Keneshea et al., 1979; Zadorozhny, 1982; Basseur and Offermann, 1986; Murray and Plane, 2005). The atomic oxygen is controlled by both photochemical and dynamical processes (turbulence, molecular and eddy diffusions, vertical and meridional transports). During the daytime, concentrations of atomic oxygen at altitudes below 80 km are about 10^{10} cm^{-3} . When the solar zenith angle exceeds 96° , photochemical production of atomic oxygen stops and its number density drops dramatically.

Ozone is continually formed and destroyed in the mesosphere. Measurements (Zommerfelds et al., 1989; Fussen et al., 2000; Kaufmann et al., 2003; Polyakov et al., 2005) and theoretical investigations (Shimazaki and Laird, 1970, 1972; Thomas and Bowman, 1972; Zadorozhny, 1982; Rodrigo et al., 1986; Sandor et al., 1997) show substantial diurnal variations of ozone concentration with maximum values at nighttime. Theoretical values of ozone numerical density are rather close to experimental values. It should be noted that differences between maximum and minimum values of ozone, derived in different theoretical models for the similar conditions, are not as large as for NO and O concentrations. During the nighttime, differences in theoretical estimations of both atomic oxygen and ozone concentration

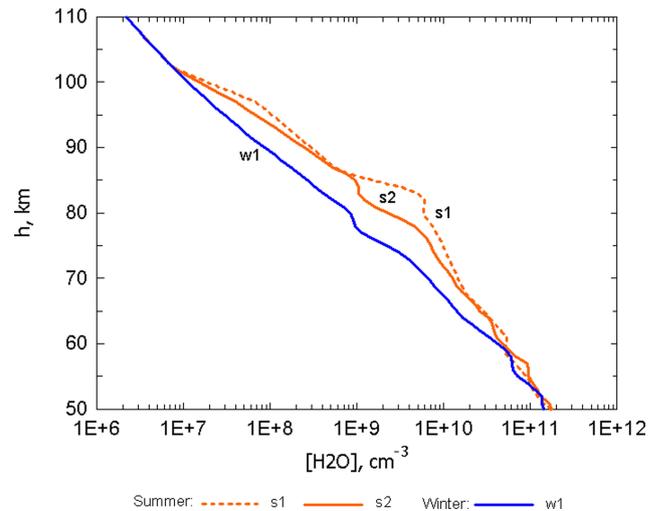


Fig. 2b. Profiles of water vapor number density (HALOE <http://haloe.gats-inc.com/home/index.php>). Curve w1 – for winter. Curves s1, s2 – for summer.

are more pronounced than during the daytime. Profiles of atomic oxygen and ozone at local noon and midnight presented in Fig. 2a are used as input parameters into the theoretical model of the D-region. The O profiles 1 and 2 at altitudes between 60 and 85 km are taken from the models by Zadorozhny (1982) and Shimazaki and Laird (1972). The O profile above 85 km is taken from the MSIS-2000 neutral atmosphere model. The O₃-3 profiles are taken from the model by Rodrigo et al. (1986). The profiles O-4 and O₃-5 at altitudes above 70 km are taken from Gumbel et al. (1998). They have been arbitrarily extrapolated downward to 60 km.

Water vapor profiles are taken from measurement by the HALOE instrument (Fig. 2b).

4 Results

To test the quality of the model and to determine which profiles of minor neutral constituents are the most appropriate for mesospheric altitudes under quiet conditions, simulated electron density profiles are compared with electron densities obtained from rocket experiments in summer and winter for solar zenith angles less than 80° (day) and larger than 99° (night). Rocket data obtained by the Faraday rotation method in the Volgograd region (geographic coordinates are 48.5° N , 45.6° E ; geomagnetic latitude is 43.1° N) have been taken from catalogues presented by Danilov and Ledomskeya (1983), Smirnova et al. (1990).

Figure 3a shows electron density profiles in the altitude range between 70 and 90 km measured near Volgograd under similar geophysical conditions, i.e. solar zenith angles of 30° – 52° , low solar activity ($F_{10.7} = 70$ – 82), summertime. The electron density profile for the altitude interval between 80 and 100 km at 37.8° N , geomagnetic latitude 49.3° N ,

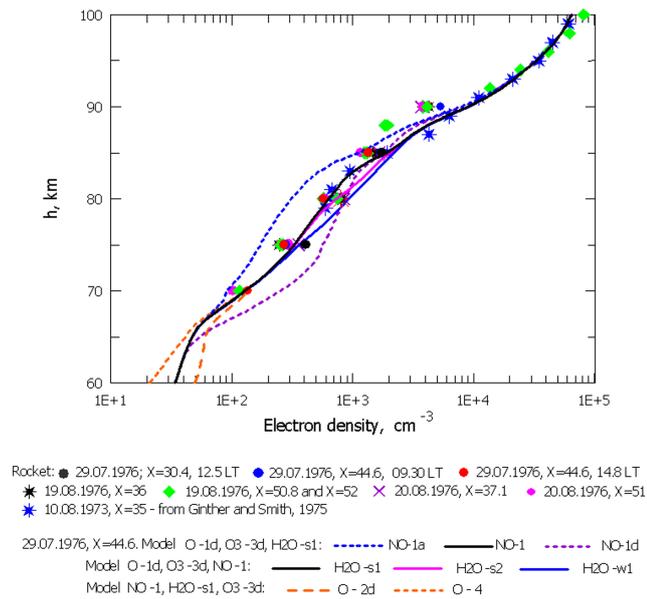


Fig. 3a. Electron number density profiles from rocket experiments (symbols) and the model (dashed and solid curves without symbols). Solar zenith angles for the rocket data 30.4°–52°. Modeled profiles are calculated for 29 July 1976, solar zenith angle 44.6°, $F_{10.7} = 70$.

solar zenith angle $\chi = 35^\circ$, has been taken from Gintner and Smith (1975). Modeled electron density profiles have been calculated for 29 July 1976, $\chi = 44.6^\circ$, $F_{10.7} = 70$ at 48.5° N, 45.6° E using all profiles of the minor neutral species presented in Figs. 1a, 2a and b. Samples of the $N_e(h)$ -profiles calculated for O-1d, O₃-3d (Fig. 2a), H₂O-s1 (Fig. 2b), but for different nitric oxide profiles (1a, 1 and 1d), are shown in Fig. 3a by the short dashed lines. It can be seen that modeled values of electron density respond to changes in the nitric oxide concentration only in the altitude interval between 68 and 85 km, since total ionization rate at these altitudes is mainly due to ionization of NO molecules by Ly- α emission (Fig. 3b). The difference between NO concentration for profiles 1a and 1d (Fig. 1a) in the altitude range 70–80 km can reach an order of magnitude. This leads to changes of electron number density of 2.8–3.5 times. At altitudes above 85 km, ionization near noon is produced mainly by other lines of the solar spectrum (Fig. 3b) and variations of NO concentration by factors 3–5 does not change the modeled values of electron density.

In Fig. 3a the influence of water vapor and atomic oxygen concentrations on the electron number density are shown. A decrease of water vapor concentration, by more than a factor of 2 at all altitudes of the mesosphere (H₂O-s1 and H₂O-w1 in Fig. 2b) leads to an increase of electron number density only in the height range between 73 and 85 km where the water concentration changes by more than factor of 5. An influence of atomic oxygen concentration (O-2d and O-4 in Fig. 2a) on electron density is seen only at the altitudes be-

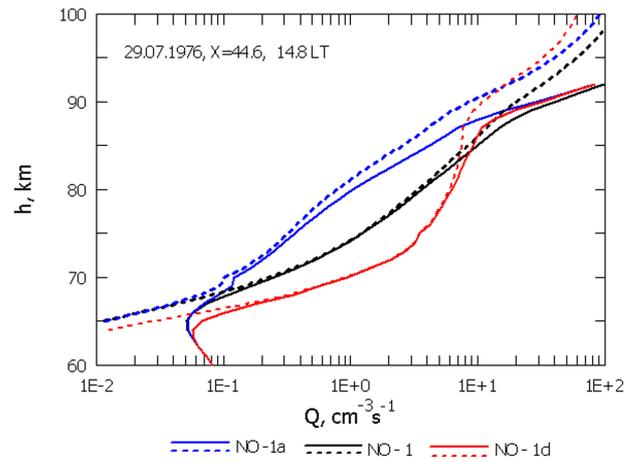


Fig. 3b. Profiles of the ionization rate of nitric oxide by Ly- α emission (dashed lines) and total ionization rate (solid lines) for 29 July 1976, solar zenith angle 44.6° on the basis of NO-1a, 1 and 1d profiles.

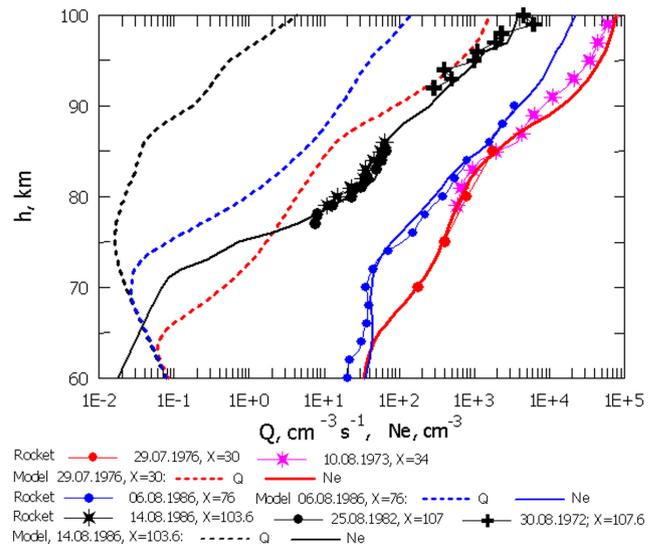


Fig. 4. Electron density profiles measured at different solar zenith angles under low solar activity $F_{10.7} = 67\text{--}85$ (symbols). Ionization rate (dashed lines) and electron number density (solid lines without symbols) modeled for the same conditions on the basis of NO-1 profile (Solomon et al., 1982).

low 70 km (red dashed lines). Above 70 km, both high (O-2d profile) and very low (O-4 profile) values of oxygen compared to O-1d profile, don't change electron number density at daytime. Changes of ozone concentration, between the O₃-3d profile and the O₃-5 profile, do not influence N_e values at altitudes above 70 km. Below 70 km, an increase of ozone by 3–4 times leads to a decrease of electron number density by less than 1.3 times. The influence of ozone on electron density is not shown in Fig. 3a. Comparing the summer modeled and experimental electron number densities, it can be

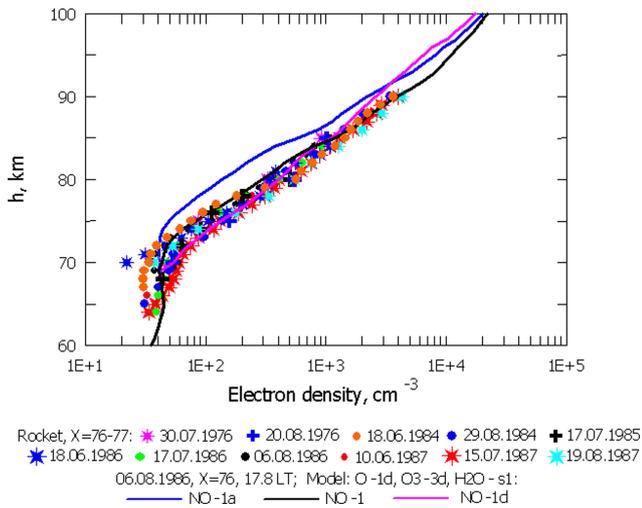


Fig. 5a. Electron number density profiles from rocket experiments (symbols) and the model (solid curves without symbols) during low solar activity. Solar zenith angles for the rocket data 76°–77°. Modeled profile is calculated for 6 August 1986, solar zenith angle 76°, $F_{10.7} = 70$ on the basis of different NO-1 profiles (solid lines) (Solomon et al., 1982).

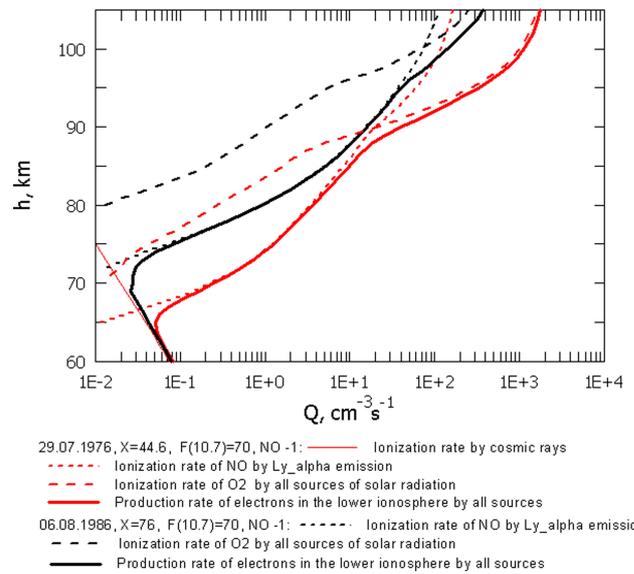


Fig. 5b. Profiles of the ionization rate for 29 July 1976, solar zenith angle 44.6° (red lines) and for 6 August 1986, solar zenith angle 76° (black lines) on the basis of the NO-1 profile (Solomon et al., 1982).

seen that just the profiles of water vapor measured in summer (the H₂O-s1 or H₂O-s2) and the daytime profile NO-1 are required in order to model electron density profiles which are consistent with observations at $\chi = 30$ –50° during low solar activity. Below, all model $N_e(h)$ -profiles for the sunlit summer ionosphere are computed on the basis of the input O-1d (Zadorozhny, 1982), O₃-3d (Rodrigo et al., 1986) and H₂O-s1 (HALOE) profiles.

The ionization rates and corresponding electron density profiles have been modeled for low solar activity at different solar zenith angles (29 July 1976, $\chi = 30.4^\circ$ and 44.6° ; 18 July 1984, $\chi = 76^\circ$; 14 August 1986, $\chi = 103.6^\circ$) as the respective experimental rocket data were available (Fig. 4). They are shown by dashed and solid lines. Experimental values of electron density are shown by symbols. It can be seen that the modeled values of the electron density reproduce experimental N_e values at all zenith angles. For daytime, the NO-1, O-1d and O₃-3d profiles have been used as input into the model. For nighttime ($\chi = 103.6^\circ$), the modeled $N_e(h)$ profile (black solid line) has been obtained on the basis of the O-1n, O₃-3n and NO-2 profiles.

In Fig. 5a eleven experimental N_e profiles for summer (June–August), measured during afternoon, at solar zenith angles of 76°–77°, and during low solar activity ($F_{10.7} = 67$ –85), are reproduced from a catalog of rocket data. The model $N_e(h)$ profiles have been calculated for 6 August 1986 at $\chi = 76^\circ$ on the basis of the NO-profiles 1a, 1 and 1d. It is noted that at this zenith angle, considerable influence of NO on electron production rate (black lines in Fig. 5b) and electron density is seen at the altitudes between 70 and 95 km.

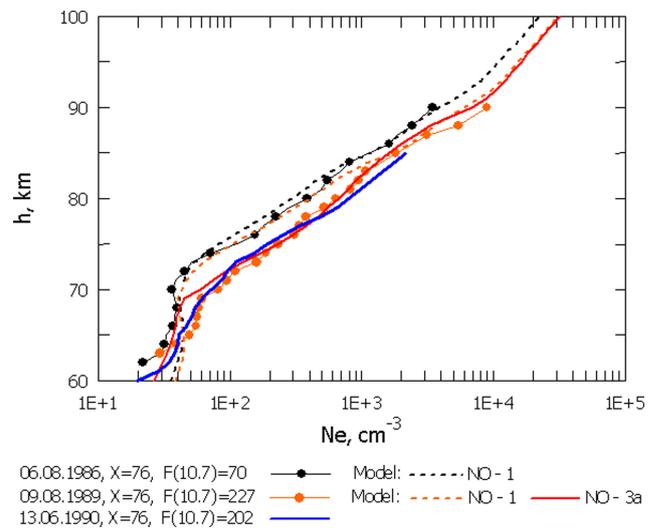


Fig. 6. Electron density profiles measured at a solar zenith angle of 76° for low and high solar activity and modeled on the basis of the NO-1 and 3a profiles for the same conditions.

Experimental data demonstrate that, under the same geographical conditions, electron number density in the lower ionosphere varies on a day-to-day basis. Differences between summer N_e values can reach about a factor 2. The model is able to explain such deviations at altitudes between 70 and 90 km by variations of NO concentration which are limited by the NO-1 and the NO-1d profiles (Solomon et al., 1982).

Electron density profiles measured after noon, at solar zenith angle 76°, but for different solar activity

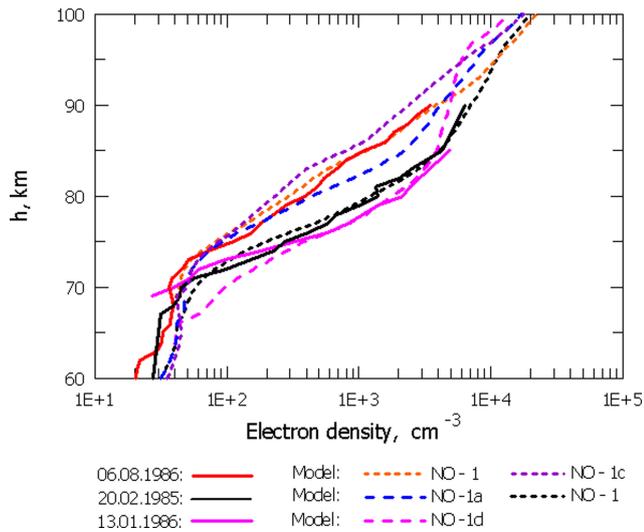


Fig. 7. Electron number density profiles measured in summer (6 August 1986, solar zenith angle 76° , $F_{10.7} = 70$) and in winter (20 February 1985 and 13 January 1986, solar zenith angle 76° , $F_{10.7} = 70$ and 76.8). Electron densities modeled on the basis of the H₂O-s1 (summer), H₂O-w1 (winter) and NO-1a, 1, 1c and 1d profiles (dashed lines).

(6 August 1986, $F_{10.7} = 70$; 9 August 1989, $F_{10.7} = 227$; 13 June 1990, $F_{10.7} = 202$) are shown in Fig. 6 with solid black, red and blue lines. Electron number densities differ from each other. For minimum and maximum activity, the difference between N_e values at altitudes 68–82 km is about a factor 2. Samples of modeled N_e profiles calculated for corresponding conditions on the basis of the daytime NO-1 profile are shown with short dashed black (low activity) and short dashed red (high activity) lines. For low solar activity, the modeled and experimental N_e profiles are in rather good agreement. The maximum deviations between them, of a factor of 1.4–1.3, are seen at altitudes between 75 and 82 km. In order to reproduce experimental data between 70 and 80 km during high solar activity, higher NO concentrations between 70 and 80 km are required than during low solar activity, in agreement with both theoretical models of NO (e.g. Shimazaki and Laird, 1972) and experimental data (HALOE).

Figure 7 shows how experimental and modeled electron density profiles respond to season changes. The electron density profiles have been measured after noon, under similar geophysical conditions but in different season, i.e. in summer (6 August 1986, $\chi = 76^\circ$, $F_{10.7} = 70$) and in winter (20 February 1985, $\chi = 76^\circ$, $F_{10.7} = 76.7$ and 13 January 1986, $\chi = 76^\circ$, $F_{10.7} = 76.8$). Seasonal differences in the electron density are rather pronounced. Electron density measured on 20 February 1985 exceeds summer values by a factor of 2.4 at 75 km and by a factor of 3.6 at 80 km. For both days, modeled electron density profiles have been calculated for all NO profiles represented in Fig. 1a. For the summer ionosphere the water vapor profile H₂O-s1 measured in sum-

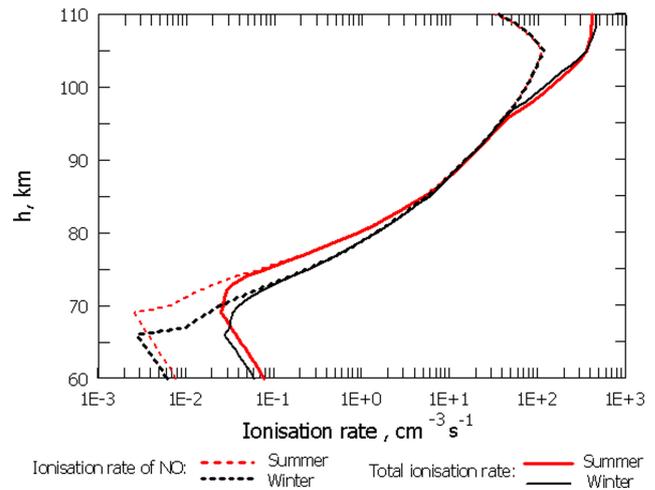


Fig. 8. Profiles of the ionization rate of nitric oxide by Ly- α emission (dashed lines) and total ionization rate for 6 August 1986 and 20 February 1985 (solid lines) on the basis of NO-1 profile.

mer is used; for the winter ionosphere – the H₂O-w1 profile. In the both cases, the best agreement between modeled and experimental data is obtained for the NO-1 profile.

In Fig. 8 quantitative estimations of the ionization rates calculated for summer and winter on the basis of the NO-1 profile are shown. Ionization rate profiles coincide at altitudes above 80 km and have small differences below 80 km. It is evident that ionization rates are not the reason for the large discrepancy between summer and winter modeled N_e profiles. The model describes seasonal differences in experimental values of electron density via meteorological factors. The efficiencies of the hydration chains of NO⁺ and O₂⁺ using water vapor profiles H₂O-s1 and H₂O-w1 are shown in Fig. 9a and b. In winter the efficiency of both hydration channels, $B(\text{NO}^+)$ and $B(\text{O}_2^+)$, is significantly less than in summer. Because of the increase of mesospheric temperature in winter (the MSIS model gives $\Delta T = 15\text{--}38$ K in the range 75–90 km), the absolute values of $B(\text{O}_2^+)$ at altitudes of about 80 km are 4 times less in winter than in summer. For $B(\text{NO}^+)$ the difference between winter and summer values is up to 13 times. Seasonal variations in the water vapor have additional effects on the efficiencies of cluster formation and the electron number density. The joint effect of seasonal changes in temperature and water vapor leads to an increase of the electron number density by up to 2.4 times at 75 km and up to 4 times at 80 km during winter, i.e. to the same seasonal differences in the model values of electron density which are observed in experimental data. Seasonal variations of effective recombination coefficient are shown in Fig. 10 (black and red solid curves).

The electron density measured on 13 January 1986 exceeds the summer N_e values by a factor of 2.9 at 75 km and of 5.5 at 80 km. The model is not able to describe such large differences in electron density by regular seasonal changes

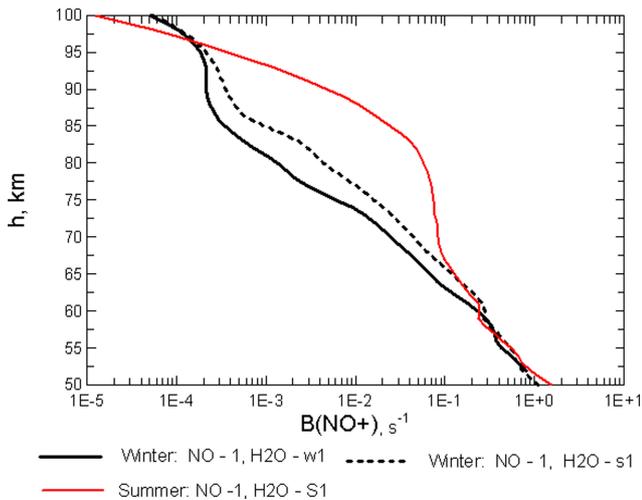


Fig. 9a. Efficiency of the hydration chain of NO^+ in summer (red line) and winter (black solid and dashed lines). Calculations are made on the basis of NO-1 profile.

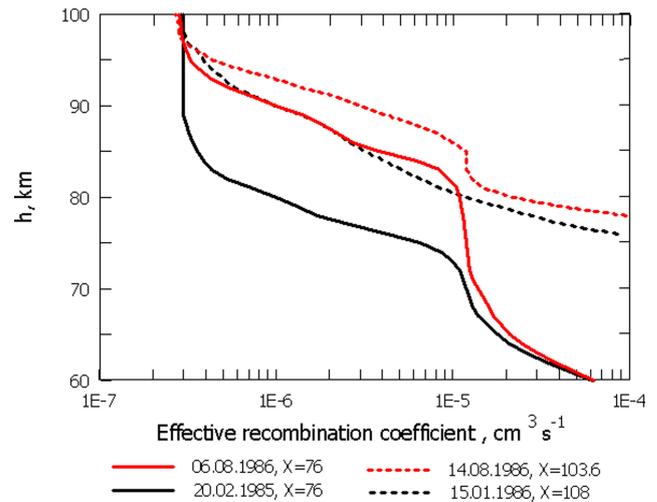


Fig. 10. Height profiles of the effective recombination coefficient calculated for summer, 6 August 1986 (red lines), and for winter, 20 February 1985 (back lines), at solar zenith angles 76° (solid lines) and larger than 100° (dashed lines).

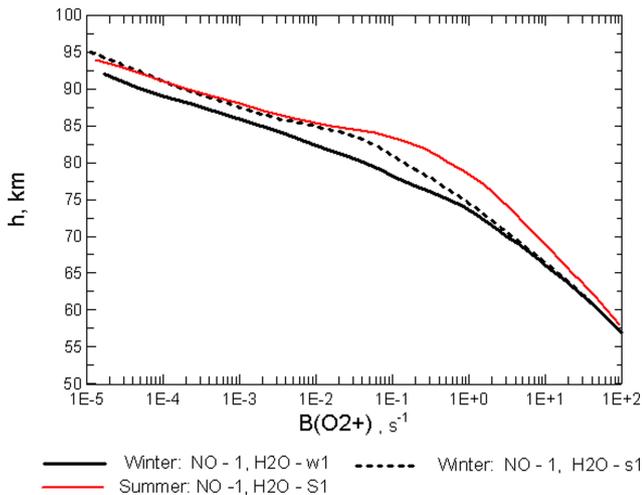


Fig. 9b. Efficiency of the hydration chain of O_2^+ in summer (red solid line) and winter (black lines).

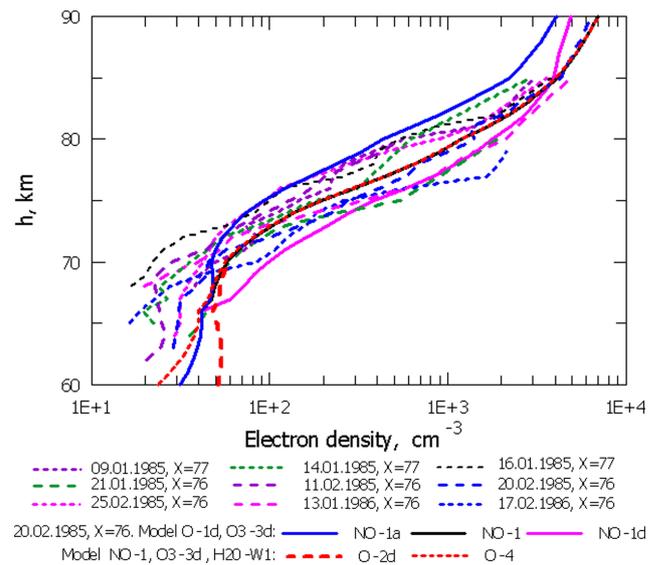


Fig. 11. Electron number density profiles measured in winter at solar zenith angles $76\text{--}77^\circ$, $F_{10.7} = 67\text{--}85$ (dashed lines). Electron density profiles modeled for 20 February 1985 on the basis of the NO-1a, 1 and 1d profiles (solid lines).

of the water vapor concentration and the neutral atmosphere temperature. In such cases higher concentrations of nitric oxide are required in order to produce additional electron density.

Figure 11 demonstrates variability in electron number density from day to day in the winter ionosphere. It is much more pronounced than in summer (Fig. 5a). The model calculations show that it might be caused by variations of NO concentration (3 solid curves in Fig. 11). It should be noted that, in winter, the required concentrations of NO lie inside a wider range of values, between the NO profiles 1a and 1d, compared to summer NO concentrations. On the other hand, it is known that, apart from regular seasonal changes of the neutral atmosphere temperature, the mesosphere is also

characterized by temperature variations during each season, due to gravity and planetary waves, atmospheric tides and other factors (Hauchecorne et al., 1987; Aikin et al., 1991; Yu et al., 1997; States and Gardner, 1998; Shepherd et al., 2001). Significant (up to ± 30 K) and irregular temperature fluctuations are observed often especially in the winter mesosphere (Theon et al., 1967; Shepherd et al., 2001). This phenomenon is known as the winter anomaly (Aikin et al., 1977; Offermann, 1979). They might also be responsible for large

variability of electron number density on winter days. Since individual temperature profiles can significantly differ from the mean temperature profiles provided by the MSIS model, the role of each factor (NO or ΔT) in formation of N_e profiles in the daytime winter ionosphere can be understood only by detailed studies using real temperature profiles of the winter mesosphere. Influence of atomic oxygen concentration on the electron number density in the winter ionosphere is shown by two dashed red lines.

Figure 12 shows electron density profiles measured in the winter ionosphere at solar zenith angles in the range of $100\text{--}110^\circ$ during low solar activity. Calculations of the modeled electron density profiles have been made on 15 January 1986, ($\chi = 108^\circ$, $F_{10.7} = 77.7$), using different night profiles of atomic oxygen, ozone and nitric oxide. The modeled profile that is very close to the experimental values has been obtained on the basis of the O-1n, O₃-3n and NO-2 profiles. This modeled $N_e(h)$ profile and the $N_e(h)$ profile calculated for 14 August 1986 at $\chi = 103.6^\circ$ (Fig. 4), have been used to determine the effective recombination coefficients at nighttime in the winter and summer ionosphere (black and red dashed curves in Fig. 10). It can be seen, that the model provides strong dependence of effective recombination coefficient on both season and solar zenith angle.

5 Discussion and conclusion

Modeled ionization rates and electron number density, based on different profiles of minor neutral constituents (NO, H₂O, O and O₃), covering a broad range of values, have been presented in this study. Correct estimation of the role of the nitric oxide concentration in formation of the electron density profile depends, first of all, on accurate calculations of the ionization rate. In Fig. 5b the ionization rates provided by different solar sources at zenith angles 44.6° and 76° ; moreover, cosmic rays are shown. The model calculations confirm that NO molecules ionized by Ly- α emission are the major source of electrons and positive ions in the quiet sunlit D-region. The height interval where the ionization rate produced by Ly- α is dominant depends on both the NO concentration (Fig. 3b) and solar zenith angle (Fig. 5b). Near noon ($\chi = 44.6^\circ$), the contribution of the Ly- α emission to the electron production rate is the largest between 68 and 85 km. Calculations using a wide spectrum of NO concentrations as model input, from minimum (the NO-1a profile) to maximum (the NO-profile 1d or 3a) values, demonstrate that the difference between the minimum and maximum values of the ionization rate at 68–85 km might be up to a factor 10 (Fig. 3b). Differences between the maximum and minimum values in the electron number density caused by NO variations are of the order of 2.8–3.5 times (Fig. 3a). When the solar zenith angle increases ($\chi = 76^\circ$) and the contribution of the Ly- α emission to the electron production rate becomes significant over a wider height range (Fig. 5b), the influence

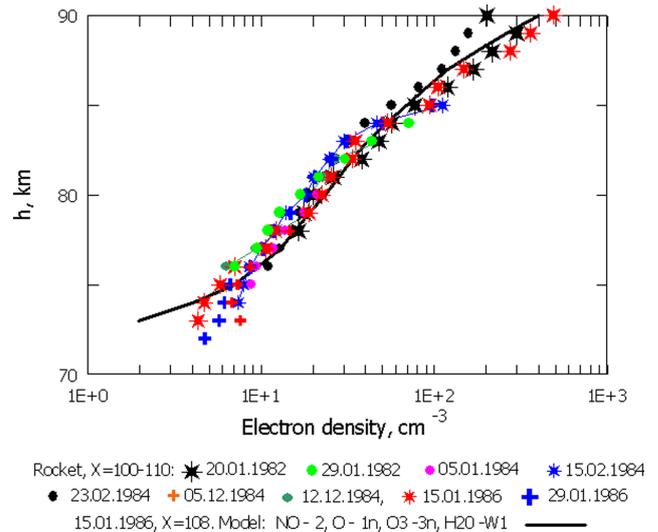


Fig. 12. Electron number density profiles measured in winter at solar zenith angles $100^\circ\text{--}110^\circ$, $F_{10.7} = 67\text{--}85$ (lines with symbols). Electron density modeled for 15 January 1986 at solar zenith angle 108° , $F_{10.7} = 77.7$ (black solid line).

of NO concentration on electron number density extends to 100 km (Fig. 5a).

It has been demonstrated that if water vapor concentration at all altitudes of the mesosphere decreases by more than a factor 2, significant increase of electron number density takes place only in the height interval between 73 and 85 km and then only if the water concentration at these altitudes changes by more than a factor 5. In the sunlit ionosphere, influence of atomic oxygen and ozone concentrations on electron density is seen only at altitudes below 70 km. A change in the atomic oxygen concentration at 65 and 60 km by a factor of 10 leads to changes in the electron density by a factor of 1.5 and 2.3, respectively. The effect of the change of ozone concentration is less significant. The model describes this altitude-dependent response of electron density to changes of water, atomic oxygen and oxygen in the frame of the ion-chemical model through the participation of these species in the ion-molecular reactions.

Comparison between experimental and model electron density profiles, calculated in each individual case under the same conditions (date, time, latitude, longitude, χ , $F_{10.7}$), demonstrates that the model is able to reproduce the electron density profiles in the quiet summer and winter ionosphere on the basis of NO profiles which are inside the range of available NO values. On summer days, during low solar activity, the profile NO-1 provides coincidence between model and experimental data for most of the cases considered. In order to keep agreement between the modeled and measured electron density profiles during high solar activity ($F_{10.7} = 200\text{--}220$) the number density of nitric oxide below 80 km should be higher than during low ($F_{10.7} = 70$) activity

by 3–4 times (Fig. 6). This corresponds to both statistical data on NO concentration in the upper mesosphere obtained from the satellite measurements for different levels of the solar activity and the conclusions deduced from diffusion photochemical models. The model calculation shows that deviations (by a factor 2) in experimental summer N_e values at altitudes 70–90 km can be stipulated by changes of NO concentration which are limited by the NO-1 and the NO-1d profiles.

Large variability in experimental winter N_e values can be caused both by strong variability of NO concentration and strong fluctuations of the temperature in the winter mesosphere. Available NO profiles (between the NO-1a and the NO-1d) might explain deviations in experimental values of electron density above 70 km (Fig. 11). On other hand, because of the strong inverse dependence of the NO⁺ hydration channel efficiency on temperature, frequent and large temperature oscillations in the winter mesosphere might be also responsible for observed deviations in electron density. The real temperature profiles can significantly differ from the MSIS profiles, especially in the region of the mesopause. Positive fluctuations of temperature decrease the cluster ions formation rate and effective recombination coefficient, and the electron density can increase by several times, depending on the altitude, compared to an average winter day, and vice versa. In order to understand the role of each factor (NO or ΔT) in formation of the N_e profile, it is necessary to incorporate the real temperature profiles into the model. In this case the possibility appears to determine which height profiles of nitric oxide are the most appropriate for the mesosphere on winter days.

Due to realistic assumptions concerning the dependence of the main cluster ions formation processes on temperature, the model is able to explain regular seasonal variations in the electron number density and effective recombination coefficient (Figs. 7 and 10) in terms of temperature and water vapor seasonal variations.

The theoretical model for quiet lower ionosphere has been applied to compute ionization rate and electron density profiles in the summer and winter D-region. In order to minimize possible errors in estimation of ionization rates by solar electromagnetic radiation and obtain the best values of electron density, each wavelength range of the solar spectrum has been divided into several intervals and the relation between the solar radiation intensity at these wavelengths and the solar activity index $F_{10.7}$, deduced by Nusinov (1992), has been implemented. The lack of reliable data on the height distribution of minor neutral constituents in the mesosphere is also a problem for reasonable estimation of the ionospheric parameters in the quiet ionosphere. In this study we have examined the influence of minor neutral species concentration on the ionization balance at different altitudes of the D-region and the response of the modeled $N_e(h)$ profiles to changes in solar zenith angle, solar activity and season. Calculations have been made at solar zenith angles less than 80° and more than

99°, under steady state conditions. Modeled electron density profiles have been evaluated by comparison with experimental profiles available from the rocket measurements made in Volgograd region (48.5° N, 45.6° E).

We have shown that our theoretical model for the quiet lower ionosphere is quite effective in describing variations in ionization rate, effective recombination coefficient and electron number density as functions of solar zenith angle, solar activity and season (for T and H₂O). The profiles O-1d or O-2d, and O₃-3d (Fig. 2a) can be used to model electron density in the sunlit ionosphere. The profiles H₂O-s1, s2, and H₂O-w1 (Fig. 2b) are suitable for modeling in the summer and winter ionosphere, respectively. The NO-1 profile (Solomon et al., 1982) gives the best values of electron density in the daytime summer ionosphere during low solar activity and the NO profiles with higher NO concentrations give the best values during high solar activity. For modeling in the daytime winter ionosphere the range of applied NO profiles is rather wide. The modeling of the ionospheric parameters on winter days is less successful than for summer conditions.

The results presented in this study are based on a theoretical model for the lower ionosphere which is relatively simple compared to the models on a planetary scale which include several hundred chemical reactions, dozens of ions, electrodynamics of plasma, neutral constituents chemistry, plasma and neutral wind dynamics, optics and etc. (Kull et al., 1997; Verronen et al., 2002; Kazil et al., 2003; Enell et al., 2011). The complex models are extremely important for detailed research of ionosphere and magnetosphere physics. Along with their undeniable merits, they have also intrinsic limitations. First of all, the limitations may be associated with inadequacies in the reaction rate constants and the uncertainties of some chemical reaction paths included in the detailed ion and neutral chemical schemes. One of the goals for construction of the “simple” models is to avoid such problems by means of reasonable simplification of the ion-molecular processes scheme. In addition, the complete and simple models may fulfill different tasks. For example, the simple model having incorporated the appropriate minor neutral constituents profiles (NO, O, O₃, H₂O, O₂(1 Δ_g), NO₂, CO₂) will allow more accurate modeling of electron and ion densities in a given geophysical situation, without the need to apply complete model calculations of all neutral species. In addition, the simple model may be applied to solve the inverse tasks. It may provide the kernel for spectra recovering methods and techniques, for example, deducing flux-energy spectrum of electrons precipitating into the polar ionosphere from ground-based or rocket measurements of electron density profile (Osepian and Kirkwood, 1996; Osepian and Smirnova, 1997). The model can be used for estimations of minor neutral species concentration in the mesosphere under quiet or disturbed conditions on basis of available experimental electron density profiles. This type of model can provide the quick information needed for calculation and forecasting of radio waves propagation in a wide range of frequencies in various

Table A1. Rate constants for the positive ion reactions, $\text{cm}^3 \text{s}^{-1}$ (Smirnova et al., 1988).

r_1	$1.8 \times 10^{-28} (300/T)^{4.7}$	K_2	3.0×10^{-10}
r_2	$2.0 \times 10^{-31} (300/T)^{4.4}$	K_3	1.0×10^{-10}
r_{-2}	$1.5 \times 10^6 T^{-5.4} \exp(-2450/T)$	K_4	2.2×10^{-9}
r_3	1.0×10^{-9}	k_5	4.4×10^{-10}
r_4	1.0×10^{-9}	k_6	1.0×10^{-17}
r_5	$7.0 \times 10^{-30} (300/T)^3$	ν_1	$8.0 \times 10^{-31} (300/T)^{4.4}$
r_{-5}	$3.1 \times 10^4 T^{-4} \exp(-4590/T)$	$\alpha(\text{O}_2^+)$	$2.0 \times 10^{-7} (300/T)$
r_6	1.0×10^{-9}	$\alpha(\text{NO}^+)$	$4.0 \times 10^{-7} (300/T)$
r_7	$1.0 \times 10^{-27} (300/T)^{4.4}$	$\alpha(\text{Cl}_1^+)$	2.0×10^{-6}
K_1	$2.4 \times 10^{-30} (300/T)^{3.2}$	$\alpha(\text{Cl}_2^+)$	1.0×10^{-5}
K_{-1}	$1.8 \times 10^{-5} (300/T)^{4.2} \exp(-5600/T)$		

Table A2. Rate constants for the negative ion reactions, $\text{cm}^3 \text{s}^{-1}$ (Smirnova et al., 1988).

β_1	$1.4 \times 10^{-29} (300/T) \cdot \exp(-600/T)$	β_{12}	$4.0 \times 10^{-31} (300/T)^5$
β_2	1.0×10^{-31}	β_{13}	$2.0 \times 10^{-29} (300/T)^5$
β_3	1.5×10^{-10}	β_{14}	6.0×10^{-10}
β_4	$2.0 \cdot 10^{-10}$	β_{15}	$9.0 \times 10^{-31} (300/T)^5$
β_5	$9.1 \times 10^{-12} (300/T)^{-1.46}$	β_{16}	$3.1 \times 10^{-28} (300/T)^5$
β_6	1.9×10^{-10}	β_{17}	5.3×10^{-10}
β_7	2.0×10^{-10}	ρ_1	0.33
β_8	1.1×10^{-10}	ρ_2	1.4
β_9	0.15	ρ_3	1.5×10^{-3}
β_{10}	1.1×10^{-11}	α_i	$6.8 \cdot T^{-0.4}$
β_{11}	2.0×10^{-10}		

geophysical conditions, i.e. they can be used for any certain practical purpose.

Appendix A

Ion chemistry model

A block diagram of the ion-chemical model of the D-region is illustrated by Smirnova et al. (1988) in Fig. 1. Equations for computing reaction rates for positive and negative ions are given below:

$$d[\text{O}_2^+]/dt = Q(\text{O}_2^+) - (k_5[\text{NO}] + k_6[\text{N}_2] + B(\text{O}_2^+) + \alpha(\text{O}_2^+) \cdot N_e + \alpha_i \cdot N^-) \cdot [\text{O}_2^+]$$

$$d[\text{NO}^+]/dt = Q(\text{NO}^+) + (k_5[\text{NO}] + k_6[\text{N}_2]) \cdot [\text{O}_2^+] - (B(\text{NO}^+) + \alpha(\text{NO}^+) \cdot N_e + \alpha_i \cdot N^-) \cdot [\text{NO}^+]$$

$$d[\text{Cl}_1^+]/dt = B(\text{O}_2^+) \cdot [\text{O}_2^+] + B(\text{NO}^+) \cdot [\text{NO}^+] - (r_7[\text{H}_2\text{O}][\text{N}_2] + \alpha(\text{Cl}_1^+) \cdot N_e + \alpha_i \cdot N^-) \cdot [\text{Cl}_1^+]$$

$$d[\text{Cl}_2^+]/dt = r_7[\text{H}_2\text{O}][\text{N}_2][\text{Cl}_1^+] - (\alpha(\text{Cl}_2^+) \cdot N_e + \alpha_i \cdot N^-) \cdot [\text{Cl}_2^+]$$

$$d[\text{O}_2^-]/dt = (\beta_1[\text{O}_2]^2 + \beta_2[\text{O}_2][\text{N}_2]) \cdot N_e + \beta_8[\text{O}][\text{CO}_3^-] - (B(\text{O}_2^-) + 2\beta_3[\text{O}] + \beta_4[\text{O}_2(^1\Delta_g)] + \rho_1 + \alpha_i \cdot N^+) \cdot [\text{O}_2^-]$$

$$d[\text{O}^-]/dt = \beta_5[\text{O}_3] \cdot N_e + \beta_3[\text{O}][\text{O}_2^-] + \beta_9[\text{CO}_3^-] - (\beta_6[\text{O}] + \beta_7[\text{O}_2(^1\Delta_g)] + \rho_2 + B(\text{O}^-) + \alpha_i \cdot N^+) \cdot [\text{O}^-]$$

$$d[\text{CO}_3^-]/dt = B(\text{O}_2^-) \cdot [\text{O}_2^-] + B(\text{O}^-) \cdot [\text{O}^-] - (\beta_8[\text{O}] + \beta_9 + \beta_{10}[\text{NO}] + \beta_{11}[\text{NO}_2] + \alpha_i \cdot N^+) \cdot [\text{CO}_3^-]$$

$$d[\text{NO}_3^-]/dt = (\beta_{10}[\text{NO}] + \beta_{11}[\text{NO}_2]) \cdot [\text{CO}_3^-] - (\rho_3 + \alpha_i \cdot N^+) \cdot [\text{NO}_3^-]$$

The efficiency $B(\text{NO}^+)$ of the channel $\text{NO}^+ \rightarrow \text{NO}^+ \cdot (\text{H}_2\text{O})$ is given by the following expression:

$$B(\text{NO}^+) = [\text{H}_2\text{O}] \left(r_1[\text{N}_2] + \frac{r_2[\text{N}_2]^2 r_4}{c} + \frac{r_6}{r_6[\text{H}_2\text{O}]r_{-5}[\text{N}_2]} \left(r_5[\text{CP}_2][\text{N}_2] + \frac{r_2[\text{N}_2]^2 r_5[\text{CO}_2]}{c} \right) \right) \quad (\text{A1})$$

where $c = r_{-2}[\text{N}_2] + r_3[\text{CO}_2] + r_4[\text{H}_2\text{O}]$. Reaction rate constants r_1, r_2, r_{-2}, r_5 and r_{-5} depend on temperature and are given in Table A1.

The efficiency $B(\text{O}_2^{2+})$ of the channel $\text{O}_2^{2+} \rightarrow \text{O}_2^{2+} \cdot (\text{H}_2\text{O})$ is

$$B(\text{O}_2^{2+}) = \frac{K_1[\text{O}_2]^2 + \nu_1[\text{N}_2]^2}{\frac{K_2[\text{O}] + K_3[\text{O}_2(^1\Delta_g)] + K_{-1}[\text{O}_2]}{K_4[\text{H}_2\text{O}]} + 1} \quad (\text{A2})$$

where $K_1, K_{-1}, K_2, K_3,$ and K_4 are the reaction rate constants given in Table A1. The constants K_1, K_{-1} and ν_1 depend on temperature.

Efficiencies of the channels $\text{O}_2^- \rightarrow \text{O}_3^-, \text{O}_4^-, \text{CO}_4^- \rightarrow \text{CO}_3^-$ and $\text{O}^- \rightarrow \text{O}_3^- \rightarrow \text{CO}_3^-$ are given by the expressions:

$$B(\text{O}_2^-) = \beta_{12}[\text{O}_2]^2 + \beta_{13}[\text{O}_2][\text{CO}_2] + \beta_{14}[\text{O}_3] \quad (\text{A3})$$

$$B(O^-) = \beta_{15}[O_2]^2 + \beta_{16}[O_2][CO_2] + \beta_{17}[O_3] \quad (A4)$$

where β_{12-17} are rate constants given in Table A2.

Proton hydrates $H^+ \cdot (H_2O)_n$ are formed from Cl_1^+ with rate β according to

$$\beta = 10^{-27} \cdot (300/T)^{4.4} [H_2O][N_2][Cl_1^+] \quad (A5)$$

Acknowledgements. The authors thank The Education, Audiovisual and Culture Executive Agency (EACEA) for support of this work within the frame of the Erasmus Mundus Master Course in Space Science and Technology – SpaceMaster, 2009–2013. The authors thank M. Friedrich and the second referee for their valuable comments and suggestions.

Editor-in-chief M. Pinnock and Topical Editor P.-L. Blelly thank M. Friedrich and one anonymous referee for their help in evaluating this paper.

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